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Chemistry Chapter 9 Review. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. sharknerdo. Terms in this set (10) Describe the relationship between Avogadro's number and a mole of a substance. A mole is defined as the amount of a substance in 6.022×10^{23} (Avogadro's number) particles.

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Chemistry Chapter 9 Review. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. kevinog. Terms in this set (9) Explain the concept of mole ratio as used in reaction-stoichiometry. What is the source of this value? It is the conversion factor that allows you to convert between any two substances. The balanced reaction

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(a) A straight horizontal line at 1.0; (b) When real gases are at low pressures and high temperatures, they behave close enough to ideal gases that they are approximated as such; however, in some cases, we see that at a high pressure and temperature, the ideal gas approximation breaks down and is significantly different from the pressure calculated by the ideal gas equation.

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